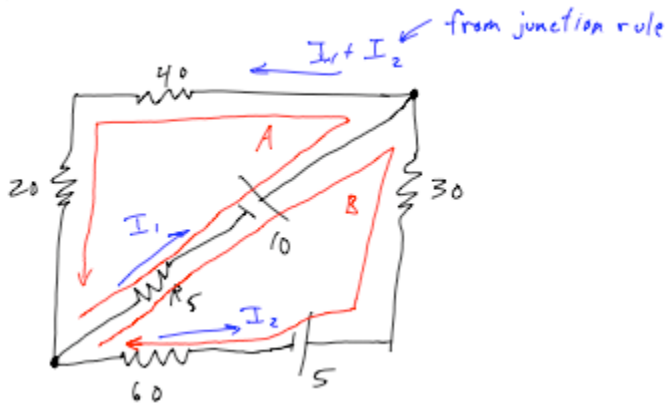


E-Bowl 2010: Preliminary Exam Answers and Solutions

1. C
2. C
3. C
4. B
5. D
6. C
7. C
8. D
9. E
10. D
11. C
12. D
13. C
14. A
15. A
16. B
17. A
18. A
19. B
20. C
21. C
22. C
23. D
24. B
25. A
26. C
27. A
28. D
29. B
30. C
31. A
32. A
33. A
34. A
35. B
36. A
37. A
38. A
39. A
40. D

1. C. Assume a composition with 100 grams. $75.69 \text{ g C}/12.0 \text{ g} = 6.3075 \text{ mol C}$;
 $8.80 \text{ g H}/1.0 \text{ g} = 8.80 \text{ mol H}$; $15.51 \text{ g O}/16 \text{ g} = 0.96375 \text{ mol O}$.
 Guess there is one mole of oxygen: $1/0.96375 = 1.031592521$
 Multiply moles of C and H by 1.031592521 to yield: 6.507 mol C and 9.078 mol H and 1 mol O
 To obtain integers, multiply by two to give: $\text{C}_{13}\text{H}_{18}\text{O}_2$
2. C. The force from m_2 is 4 times as great as the force from m_1 . To balance, the test mass needs to be twice as far away from m_2 as it is from m_1 . The only solution then is for the mass to be $d/3$ from m_1 and $2d/3$ from m_2 .
3. C. Note that the derivatives repeat; thus, $f'' = f^{(5)}$. To find $f^{(2001)}$, we only need $f'' = \sin(x) + \cos(x)$
4. B. Potential due to point charge or charged sphere: $V = Q_0/(4 * \pi * \epsilon_0 * r)$, $r = 2R$, $V(2R) = 9 * 10^9 * 3 * 10^{-12}/(2 * 0.01) = 1.35 \text{ V}$.
5. D. Use $PV = nRT$, where $R = 0.0821 \text{ L} * \text{atm}/(\text{mol} * \text{K})$, $T = 373 \text{ K}$, $P = 1 \text{ atm}$. 64 g/mol
6. C. $f'(x) = 3x^2 + 8x + 5$. Critical values are -1 and -5/3. -1 yields a minimum and -5/3 yields a maximum. $f(-1) = -1 + 4 - 5 + k = 5$. $k = 7$
7. C. limit as $1/\cos(x) = 1$
8. D. Maximum and minimum are three standard deviations from the mean. Thus, 99.7%.
9. E. Take the second derivative of each position function to obtain an acceleration function. None of them are constant.
10. D. Impulse if $F * dt$. F is $0.2 \text{ kg} * 9.8 \text{ m/s}^2$. To find the time, use the kinematic equations. The time to fall from 30 m to 0 m is 2.47 seconds, and the time from 0 m back to 15 m is 1.75 seconds ($d = vt + 1/2at^2$). Thus, the total time is 4.22 seconds. $0.2 * 9.8 * 4.22 = 8.3 \text{ N} * \text{s}$.
11. C. After a very long time, the inductor has no effect and acts as wire. Thus, $I = V/R = 9/0.90 = 10.0 \text{ A}$.
12. D. $[(1000 \text{ g vapor}) / (18.016 \text{ g vapor/mol vapor})](40.7 \text{ kJ/mol vapor})(1 \text{ mol ice} / 6.01 \text{ kJ})(18.016 \text{ g ice/mol ice}) = 6772 \text{ g} = 6.67 \text{ kg ice}$
13. C. First derivative is $1/(2\sqrt{x})$. At $x=4$, the derivative is $1/4$.
14. A. $V = 4/3 \pi r^3$, $dV = 4 \pi r^2 dr = 3.6 \pi$

15. A.



A: $-R_5 I_1 + 10 - 40 (I_1 + I_2) - 20 (I_1 + I_2) = 0$

B: $-R_5 I_1 + 10 + 30 I_2 - 5 + 60 I_2 = 0$

Simplified: A: $(60 + R_5) I_1 + 50 I_2 = 10$; B: $R_5 I_1 - 90 I_2 = 5$

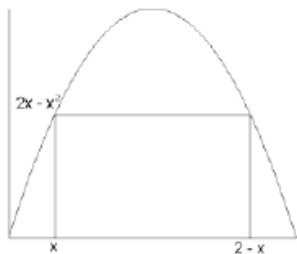
$3A + 2B: (5R_5 + 180) I_1 = 40$

$I_1 = 8 / (R_5 + 36) = 1/12$ from bottom left to top right

16. B. If you fail to reject H_0 when it is false, it is a Type II Error.

17. A. The equation is homogeneous.

18. A.



Area = $A(x) = (2 - 2x)(2x - x^2) = 2x^3 - 6x^2 + 4x$

$A'(x) = 6x^2 - 12x + 4 = 0 \quad x = 1 \pm \sqrt{3}/3$

Since $1 - \sqrt{3}/3$ is between 0 and 1, then this is the value for x

$A(1 - \sqrt{3}/3) \approx 0.77 \quad \mathbf{A}$

19. B. $0.6^{10} = 0.0060$.

20. C. Charge density = $Q_0/L = 2.5 \cdot 10^{-4} \text{ C/m}$

dq (the charge of a small piece of rod with length dx) = charge/length * length of piece = λdx

$dE = dq / (4\pi \epsilon_0 r^2) = \lambda dx / (4\pi \epsilon_0 (b+x)^2)$

$E = \text{integral of } dE \text{ from } 0 \text{ to } L = \lambda / (4\pi \epsilon_0) * (-1/(b+L) + 1/b) = 7.5 \cdot 10^6 \text{ N/C}$

21. C. In general, power is the rate of energy transfer. Here, it's the rate at which electrical energy converts into heat energy. So, the inventor wants to maximize the power. Since $P = i^2 R$, it follows that she wants to maximize the current through the resistor. In other words, she wants the peak current through the circuit, i_m , to be as large as possible.

In analogy with Ohms Law for DC circuits ($i = V/R$), the maximum current that flows through an AC circuit is $i_m = \epsilon_m / Z$, where Z is the impedance $(\sqrt{R^2 + (\omega L - 1/(\omega C))^2})$

To make Z as small as possible for a given R, the stuff inside the parentheses ($\omega L - 1/(\omega C)$), must equal zero. Therefore, $\omega = 2\pi f = 2\pi \cdot 60 \text{ Hz} = 377 \text{ s}^{-1}$. Then, $L = 1/(\omega^2 C) = 1/(377^2 \cdot 0.2 \text{ F}) = 3.52 \cdot 10^{-5} \text{ H}$.

22. C. This statement describes **Thévenin's theorem**.

23. D.

$$\frac{dy}{dx} = x + \frac{3x}{\sqrt{25-x^2}} \Rightarrow y = \frac{x^2}{2} - 3\sqrt{25-x^2} + C$$

Using the initial condition of (0, -10), $C = 5$.

Thus, the answer is $y = \frac{x^2}{2} - 3\sqrt{25-x^2} + 5$. **D**

24. B. A concentrated upward force F makes the V curve jump up by F.

25. A. $APF = N_{\text{atoms}} \cdot V_{\text{atom}} / V_{\text{unitcell}}$; $V_{\text{atom}} = 2R\sqrt{2}$; $N = 4$, $V_{\text{atom}} = 4/3 \cdot \pi \cdot R^3$; for FCC, $a = 2R\sqrt{2}$; $a^3 = 16R^3 \sqrt{2}$; $APF = 0.74$.

26. C. Three independent state properties would be required.

27. A. $F_{AB} = 30 \text{ kN}$ (Compressive); $F_{BC} = 70 \text{ kN}$ (compressive)

$$\Delta L_{AB} = FL/(AE) = 0.25 \cdot (-30,000) / (\pi \cdot 0.03^2 / 4 \cdot 200 \cdot 10^9) = -5.305 \cdot 10^{-4} \text{ m} = -0.05305 \text{ mm}$$

$$\Delta L_{BC} = 0.3 \cdot (-70,000) / (\pi \cdot 0.05^2 / 4 \cdot 105 \cdot 10^9) = -1.01859 \cdot 10^{-4} \text{ m} = -0.10185916 \text{ mm}$$

$$\Delta L_{ABC} = \Delta L_{AB} + \Delta L_{BC} = -0.1549 \text{ mm}$$

28. D. $F = \text{degrees of freedom}$, $C = \text{number of components}$, $P = \text{number of phases}$;

$$F = C - P + 2; F = 2 - 1 + 2 = 3.$$

29. B. $1 - e^{(-.25(X_a - X_b)^2)}$, where X_a and X_b are the values of electronegativity of the elements.

30. C. Net heat flow is the area of the triangle. $A = 2T_0 \cdot 4S_0 / 2 = 4T_0 S_0$

31. A. To find the strongest acid, another H^+ must be added. The only two available spots are the nitrogen and oxygen. Since nitrogen is the less electronegative of the two, $[CH_3NH_2OH]^+$ is the answer.

32. A. Markovnikov's Rule states this principle.

33. A. Inheritance states this principle.

34. A. Choice A complies with the standard for "competence."

35. B. Titles of graphs must be below the graphs.

36. A. "Recommend" is an action verb; "of" and "for" are prepositions.

37. A. Current dollars are the cash flow at the time of a transaction.

38. A. Christian Ramirez was a former Engineering Bowl director. Danielle Rickman was the executive director for E-Week 2009, and Matthew Sobieski was a former E-Fair director. David Hwang is the present E-Bowl director.

39. A. The first four departments were Mechanical, Chemical, Civil, and Electrical Engineering. Aerospace engineering was not.

40. D. When you enter Weil Hall from the west, you are on the third floor.